Pneumomediastinum and pneumorrhachis from recreational nitrous oxide inhalation: no laughing matter

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A 16-year-old girl with no medical history presented to our emergency department with rapid onset and progressive facial and neck swelling, dyspnoea and dysphonia. The previous night she had attended a nightclub, had sniffed ketamine and inhaled nitrous oxide. On examination she had swelling of her face, neck and upper torso with palpable crepitus. Scattered bilateral expiratory wheeze was present on auscultation. She was mildly tachycardic and tachypnoeic but other observations were normal. Neurological examination was normal. A chest radiograph demonstrated extensive subcutaneous emphysema. CT scan of the neck, chest and abdomen confirmed the presence of extensive subcutaneous gas within the face, neck and thorax, including pneumomediastinum, pneumopericardium and small bilateral pneumothoraces (Figure 1). Air was also seen tracking within the spinal canal in the cervical and thoracic regions, a phenomenon known as pneumorrhachis (Figure 1, arrow).

She later reported that she had taken the nitrous oxide directly from the pressurised canister (Figure 2). We felt this was likely to account for the disproportionate volume of subcutaneous and mediastinal gas. Although a pharyngeal site of perforation was thought most likely, given the distribution of subcutaneous gas and presumed localised barotrauma-induced mechanism, flexible nasendoscopy was normal.

The diffuse subcutaneous emphysema, including pneumorrhachis, was managed conservatively with supplemental oxygen therapy. There was progressive resolution over the next few days and the patient was well when reviewed at 1 month with a normal chest radiograph.

Nitrous oxide is a widely used anaesthetic agent and a recreational drug. While medical use is falling, recreational use is increasing, with 7.6% of persons aged 16–24 years having reported use in the last year,3 commonly at parties and festivals. A widely used method for recreational consumption involves decanting the contents of the pressurised canister into a balloon, presumably to allow controlled administration in small aliquots. The typical canister in Figure 2 contains 8 g of nitrous oxide, equivalent to 8 L at standard temperature and pressure. We hypothesise that our patient was experiencing the dissociative effects of the ketamine and therefore was insensitive to the localised barotrauma from the pressurised flow from the pressurised canister.

Contributors ANT wrote the manuscript and prepared the images. DL edited the manuscript and cared for the patient. SSH edited the manuscript and images. DDC edited the manuscript and

Image 1. Axial CT of the upper thorax showing extensive subcutaneous and mediastinal gas, small bilateral pneumothoraces along with pneumorrhachis (arrow).

Image 2. Typical pressurised canister in which nitrous oxide is sold for commercial and recreational use.
was clinician responsible for the patient.

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**REFERENCES**

